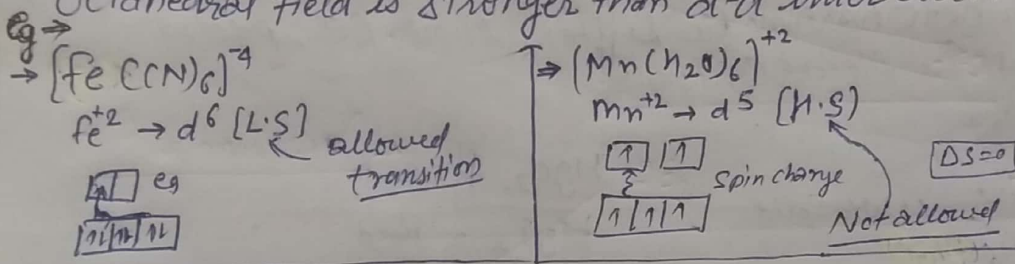


Orgel diagrams (inter electronic interaction)

Orgel Diagrams are correlation diagrams which show the relative energies of electronic terms in transition metal complexes. They are named after their creator Leslie Orgel.

- * Orgel diagrams are used for interpretation of spin allowed absorption bands of d-d origin in the electronic spectra of tetrahedral and octahedral transition metal complexes.
- * It is used to assign the electronic transition in Weak field or high spin transition metal complex and all tetrahedral complex.
- * It cannot be used for low spin or strong field complexes. because Octahedral field is stronger than d-d inter electronic repulsion.



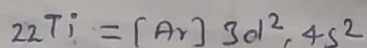
Mulliken term Symbol →

Mulliken term Symbol	Octahedral filled (Oh)	Tetrahedral filled (Td)
S	A _{1g}	A ₁
P	T _{1g}	T ₁
D	T _{2g} + E _g	T ₂ + E
F	T _{1g} + T _{2g} + A _{2g}	T ₁ + T ₂ + A ₂
G	T _{1g} + T _{2g} + E _g + A _{1g}	T ₁ + T ₂ + E + A ₁
H	2T _{1g} + T _{2g} + E _g	2T ₁ + T ₂ + E
I	2T _{2g} + T _{1g} + E _g + A _{1g} + A _{2g}	2T ₂ + T ₁ + E + A ₁ + A ₂

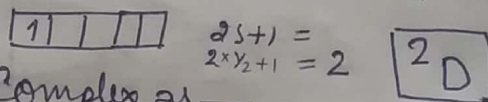
- ⊙ S=0 d₁ + l = 0 $\begin{array}{|c|} \hline \uparrow\downarrow \\ \hline \end{array}$
- P=1 l₁ + l₂ + l₃ = 1 $\begin{array}{|c|c|} \hline \uparrow & \uparrow \\ \hline \end{array}$
- d=2 d₁ + l₂ + l₃ + l₄ + l₅ $\begin{array}{|c|c|c|c|c|} \hline \uparrow & \uparrow & \uparrow & \uparrow & \uparrow \\ \hline \end{array}$
- F=3

Configuration	Energy term	Ground state & Excited state Term of same multiplicity
d^1, d^9	$2D$	$2T_{2g}, 2E_g$
d^2, d^8	$3P$ $3F$	$3T_{1g}(P)$ $3T_{1g}(F), 3T_{2g}, 3A_{2g}$
d^3, d^7	$4P$ $4F$	$4T_{1g}(P)$ $4T_{1g}(F), 4T_{2g}, 4A_{2g}$
d^4, d^6	$5D$	$5T_{2g}, 5E_g$

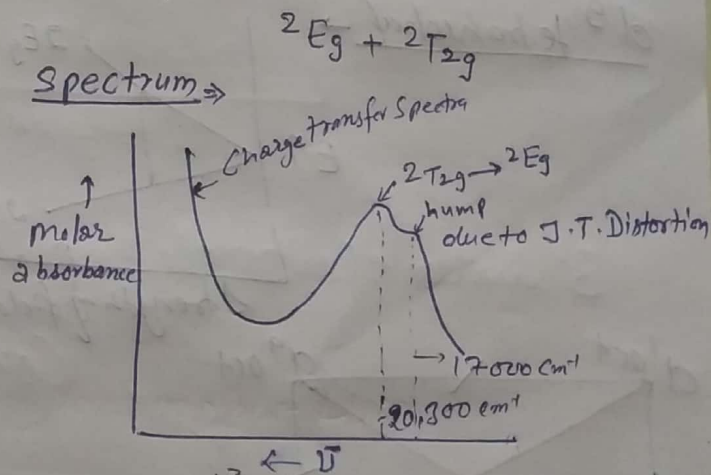
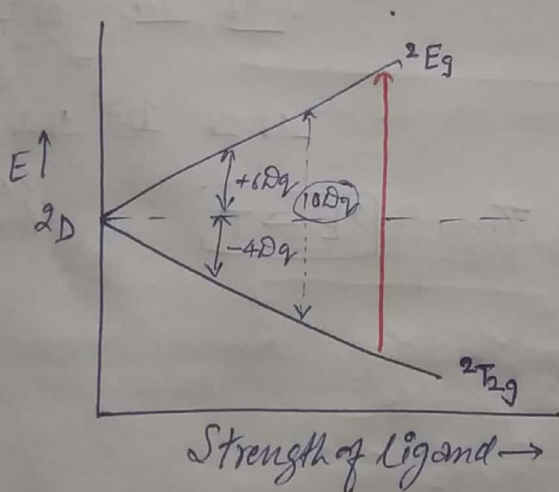
d^1 system - $[Ti(H_2O)_6]^{+3}$



$Ti^{+3} = [Ar], 3d^1$ The G.S term for d^1 sys is



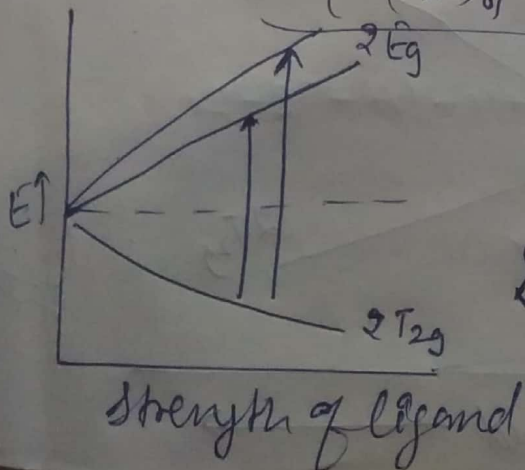
$\Rightarrow 2D$ splits in Octahedral field Complex as —



Que \rightarrow If $[Ti(H_2O)_6]^{+3} = 20,000 \text{ cm}^{-1}$

then find the value of CFSE (H.W)

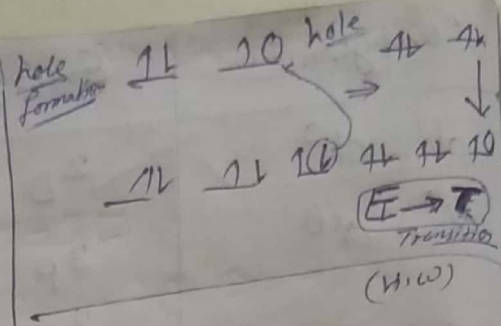
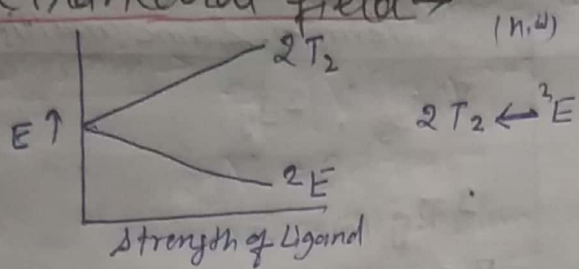
Ans $[Ti(H_2O)_6]^{+3}$ has d^1 system.



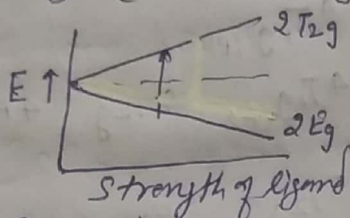
$\therefore 10 Dq = 20000 \text{ cm}^{-1}$
 $4 Dq = \frac{20000}{10} \times 4 = 8000 \text{ cm}^{-1}$

Splitting takes place due to Jahn-Teller distortion (So gives 2 peaks)

In tetrahedral field →



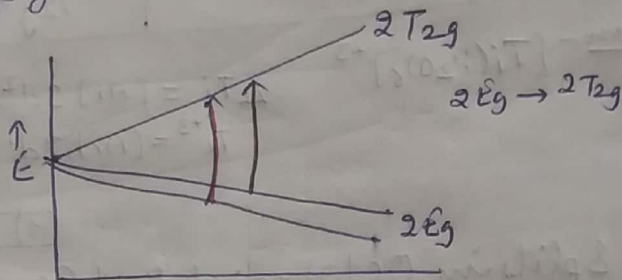
d⁹ system → eg → [Cu(N₂O)₆]⁺²
(oh)



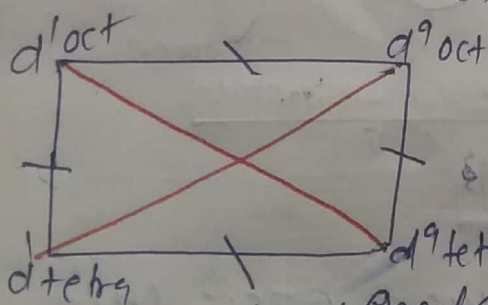
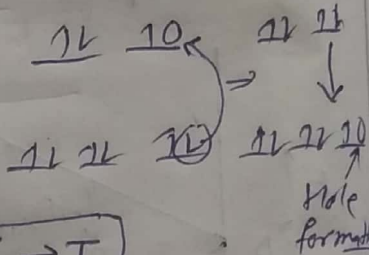
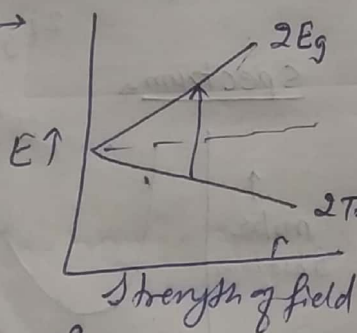
G.S Term = ²D → ²T_{2g} + ²E_g
Cu⁺² = d⁹

due to J-T. Distortion

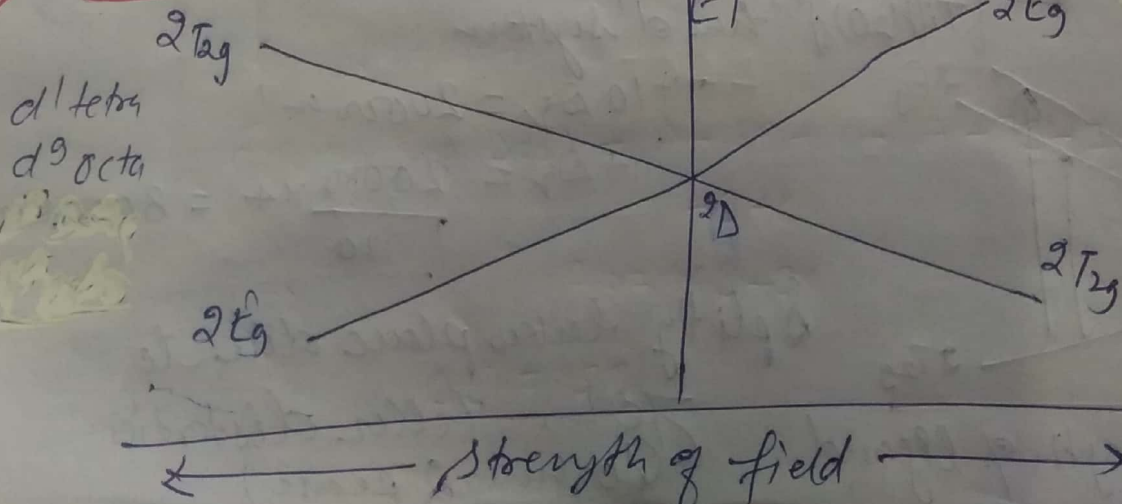
$\Delta E =$



d⁹ tetrahedral →

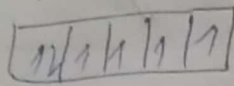


Combined Orgel diagram



(9)

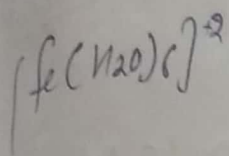
d⁶ system (CN)



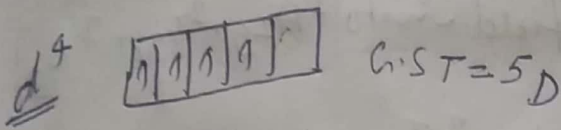
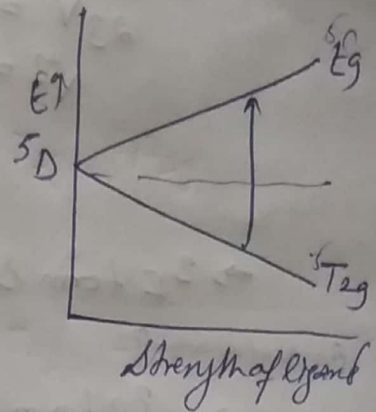
$S=2 \quad 2S+1=5$

$L=2$

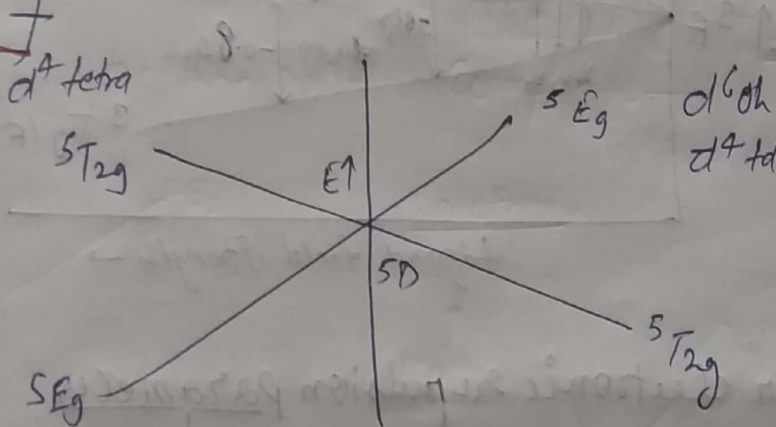
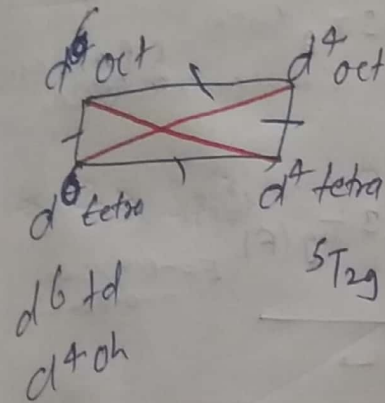
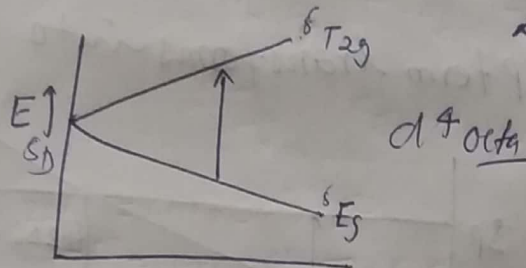
$G.S.T=5D$



$5D \rightarrow 5T_{2g} + 5E_g$



↑ 0
4 4 7
hole formulation



Combined Orgel diagram for d¹, d⁴, d⁶, d⁹ system ← Strength of Ligand →

